

# Inside the NGSS Box

## Title and Code

Two sets of performance expectations at different grade levels may use the same name if they focus on the same topic. The code, however, is a unique identifier for each set of performance expectations based on the grade level, content area, and topic of the set.

## Performance Expectations

A statement that combines practices, core ideas, and crosscutting concepts together to describe how students can show what they have learned.

## Clarification Statement

A statement that supplies examples or additional clarification to the performance expectation.

## Assessment Boundary

A statement that provides guidance about the scope of the performance expectation at a particular grade level.

## What is Assessed

A collection of several performance expectations describing what students should be able to do to master this standard

## Foundation Box

The practices, core disciplinary ideas, and crosscutting concepts from the *Framework for K-12 Science Education* that were used to form the performance expectations

## Connection Box

Other standards in the *Next Generation Science Standards* or in the *Common Core State Standards* that are related to this standard

MS.PS-SPM Structure and Properties of Matter		
<p><b>Students who demonstrate understanding can:</b></p> <p><b>a. Construct and use models to explain that atoms combine to form new substances of varying complexity in terms of the number of atoms and repeating subunits.</b> [Clarification Statement: Examples of atoms combining can include hydrogen (<math>H_2</math>) and Oxygen (<math>O_2</math>) combining to form hydrogen peroxide (<math>H_2O_2</math>) or water (<math>H_2O</math>). Assessment Boundary: Valence electrons and bonding energy are not addressed.]</p> <p><b>b. Plan investigations to generate evidence supporting the claim that one pure substance can be distinguished from another based on characteristic properties.</b> [Clarification Statement: Properties of substances can include melting and boiling points, density, solubility, reactivity, flammability, and phase.]</p> <p><b>c. Use a simulation or mechanical model to determine the effect on the temperature and motion of atoms and molecules of different substances when thermal energy is added to or removed from the substance.</b> [Assessment Boundary: Quantification of the model or use of mathematical formulas are not intended.]</p> <p><b>d. Construct an argument that explains the effect of adding or removing thermal energy to a pure substance in different phases and during a phase change in terms of atomic and molecular motion.</b> [Assessment Boundary: The use of mathematical formulas is not intended.]</p>		
<p>The performance expectations above were developed using the following elements from the NRC document <i>A Framework for K-12 Science Education</i>.</p>		
<p><b>Science and Engineering Practices</b></p> <p><b>Developing and Using Models</b> Modeling in 6-8 builds on K-5 and progresses to developing, using and revising models to explain, explore, and predict more abstract phenomena and design systems.</p> <ul style="list-style-type: none"> <li>Use and/or construct models to predict, explain, and/or collect data to test ideas about phenomena in natural or designed systems, including those representing inputs and outputs.</li> </ul> <p><b>Planning and Carrying Out Investigations</b> Planning and carrying out investigations to answer questions or test solutions to problems in 6-8 builds on K-5 experiences and progresses to include investigations that use multiple variables and provide evidence to support explanations or design solutions.</p> <ul style="list-style-type: none"> <li>Plan and carry out investigations individually and collaboratively, identifying independent and dependent variables, and controls. (b)</li> <li>Collect data and generate evidence to answer scientific questions or test design solutions under a range of conditions. (b)</li> </ul> <p><b>Engaging in Argument from Evidence</b> Engaging in argument from evidence in 6-8 builds from K-5 experiences and progresses to constructing a convincing argument that supports or refutes claims for either explanations or solutions about the natural and designed world.</p> <ul style="list-style-type: none"> <li>Use oral and written arguments supported by empirical evidence and reasoning to support or refute an explanation for a phenomenon or a solution to a problem. (d)</li> </ul>	<p><b>Disciplinary Core Ideas</b></p> <p><b>PS1.A: Structure and Properties of Matter</b></p> <ul style="list-style-type: none"> <li>All substances are made from some 100 different types of atoms, which combine with one another in various ways. Atoms form molecules that range in size from two to thousands of atoms. (a)</li> <li>Pure substances are made from a single type of atom or molecule. Each pure substance has characteristic physical and chemical properties (for any bulk quantity under given conditions) that can be used to identify it. (b)</li> <li>Gases and liquids are made of molecules or inert atoms that are moving about relative to each other. (d)</li> <li>In a liquid, the molecules are constantly in contact with others; in a gas, they are widely spaced except when they happen to collide. In a solid, atoms are closely packed and vibrate in position but do not change relative locations. (c),(d)</li> <li>Solids may be formed from molecules or they may be extended structures with repeating subunits (e.g., crystals). (a)</li> <li>The changes of state that occur with variations in temperature or pressure can be described and predicted using these models of matter. (c),(d)</li> </ul> <p><b>PS3.A: Definitions of Energy</b></p> <ul style="list-style-type: none"> <li>The term "heat" as used in everyday language refers both to thermal motion (the motion of atoms or molecules within a substance) and radiation (particularly infrared and light). (c),(d)</li> <li>Temperature is not a measure of energy; the relationship between the temperature and the total energy of a system depends on the types, states, and amounts of matter present. (c),(d)</li> </ul>	<p><b>Crosscutting Concepts</b></p> <p><b>Patterns</b> Macroscopic patterns are related to the nature of microscopic and atomic-level patterns. Patterns in natural and designed systems can provide information about natural and human designed systems. Patterns can be used to identify cause and effect relationships. Graphs and charts can be used to identify patterns in data. (c)</p> <p><b>Cause and Effect</b> Relationships can be classified as causal or correlational, and correlation does not necessarily imply causation. Cause and effect relationships may be used to predict phenomena in natural or designed systems. Phenomena may have more than one cause, and some causal relationships in systems can only be described using probability. (c),(d)</p> <p><b>Structure and Function</b> Simple and microscopic structures and systems can be visualized, modeled, and used to describe how their function depends on the shapes, composition, and relationships among its parts. Therefore complex natural and designed structures/systems can be analyzed to determine how they function. Structures can be designed to serve particular functions by taking into account properties of different materials, and how materials can be shaped and used. (b)</p>
<p><b>Connections to other DCIs in this grade-level: MS.ESS-ESP, MS.ESS-SS, MS.LS-MOE</b></p> <p><b>Articulation of DCIs across grade-levels: 3.IF, 5.SPM, HS.FS.SPM, HS.PS-NP, HS.PS-E</b></p> <p><b>Common Core State Standards Connections: (Note: these connections will be made more explicit and complete in future draft releases)</b></p> <p><b>ELA –</b></p> <p><b>W.5.2</b> Write informative/explanatory texts to examine a topic and convey ideas and information clearly.</p> <p><b>W.6.1</b> Write arguments to support claims with clear reasons and relevant evidence.</p> <p><b>W.7.1</b> Write arguments to support claims with clear reasons and relevant evidence.</p> <p><b>SL.5.4</b> Report on a topic or text or present an opinion, sequencing ideas logically and using appropriate facts and relevant, descriptive details to support main ideas or themes; speak clearly at an understandable pace.</p> <p><b>SL.6.4</b> Present claims and findings, sequencing ideas logically and using pertinent descriptions, facts, and details to accentuate main ideas or themes; use appropriate eye contact, adequate volume, and clear pronunciation.</p> <p><b>SL.7.4</b> Present claims and findings, emphasizing salient points in a focused, coherent manner with pertinent descriptions, facts, details, and examples; use appropriate eye contact, adequate volume, and clear pronunciation.</p> <p><b>WHST.6-8.1</b> Write arguments focused on discipline-specific content.</p> <p><b>RST.6-8.3</b> Follow precisely a multistep procedure when carrying out experiments, taking measurements, or performing technical tasks.</p> <p><b>Mathematics –</b></p> <p><b>MP.4</b> Model with mathematics.</p> <p><b>MP.8</b> Look for and express regularity in repeated reasoning.</p> <p><b>6.SP</b> Develop understanding of statistical variability Summarize and describe distributions</p>		

## Lowercase Letters

Lowercase letters at the end of practices, core ideas, and crosscutting Concepts designate which Performance expectation incorporates them.

## Scientific & Engineering Practices

Activities that scientists and engineers engage in to either understand the world or solve a problem

## Disciplinary Core Ideas

Concepts in science and engineering that have broad importance within and across disciplines as well as relevance in people's lives.

## Crosscutting Concepts

Ideas, such as *Patterns* and *Cause and Effect*, which are not specific to any one discipline but cut across them all.